# EFFECT OF $\mathrm{Ta}_{2} \mathrm{O}_{5}$ DOPING ON THE MICROSTRUCTURE AND DIELECTRIC PROPERTIES OF $\mathrm{BaTiO}_{3}$ BASED CERAMICS 

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#### Abstract

The main focus of the research was to correlate composition and sintering parameters with the microstructure and dielectric properties of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ ceramics. The samples were sintered using both single and two-stage sintering techniques. Thereafter SEM and XRD techniques were used to examine the structure of the samples with a particular focus on the incorporation of $\mathrm{Ta}^{5+}$ ions into the $\mathrm{BaTiO}_{3}$ crystal lattice. The SEM analysis focused on measuring the grain size and investigating the grain size distribution of the sintered samples. Finally, the dielectric properties were analysed and the relationship between the properties and structure of the doped $\mathrm{BaTiO}_{3}$ was established. From the research it can be stated that two-stage sintering yielded the best dielectric properties. The best stable value of the room temperature dielectric constant $(k)$ of 19000 was obtained for the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 6 hrs , due to the combination of a high percent theoretical density ( $\% T D$ ) and optimum grain size. At a temperature range of $30^{\circ}$ to $60^{\circ} \mathrm{C}$, this combination of composition and sintering parameters yielded a dielectric constant in the range of 18000-19000.


Keywords: Barium titanate; tantalaum oxide; doping; ceramic; double-stage sintering.

## INTRODUCTION

$\mathrm{BaTiO}_{3}$, a ferroelectric ceramic material, has attracted considerable interest for application in a variety of fields, such as capacitors, transducers, actuators, electro-optic devices. There are several reasons for its extensive practical application. Firstly because it is chemically and mechanically very stable; secondly it exhibits ferroelectric properties at and above room temperature, and finally because it can be easily prepared and used in the form of ceramic polycrystalline samples. However, pure $\mathrm{BaTiO}_{3}$ with an average grain size of around $1 \mu \mathrm{~m}$ exhibits a dielectric constant of 3000-5000 at room temperature (Amarande et al., 2007; Arlt et al., 1985; Sivakumar et al., 2011; Jingkun et al., 1999). Dielectric constants as high as 5800 have also been reported (Amarande et al., 2007). For pure $\mathrm{BaTiO}_{3}$, the dielectric constant value increases with a decrease in grain size and a general broadening of the transition peak results (Kinoshinta and Yamaji, 1976). Moreover, the suppression of the transition peak significantly increases the room temperature dielectric constant of pure $\mathrm{BaTiO}_{3}$ (Burfoot and Martirena, 1974). Despite all these findings, the industrial application of pure $\mathrm{BaTiO}_{3}$ is still somewhat limited due to its low dielectric properties.

In order to use $\mathrm{BaTiO}_{3}$ on an industrial scale, $\mathrm{BaTiO}_{3}$ has been doped with several dopants like $\mathrm{MgO}, \mathrm{ZrO}_{2}, \mathrm{Nd}_{2} \mathrm{O}_{3}, \mathrm{Ta}_{2} \mathrm{O}_{5}, \mathrm{Nb}_{2} \mathrm{O}_{5}$, and many more to control its grain size
and to improve its dielectric properties (Armstrong et al., 1989; Kelvin et al., 2010; Chan et al., 1986; Chao et al., 2008; Hwang et al., 2000; Kelvin et al., 2010; Mastelaro et al., 2004). Among these dopants, pentavalent oxides such as $\mathrm{Ta}_{2} \mathrm{O}_{5}, \mathrm{Nb}_{2} \mathrm{O}_{5}$ have a strong influence on the dielectric properties of $\mathrm{BaTiO}_{3}$. They influence the dielectric properties of pure $\mathrm{BaTiO}_{3}$ by controlling the grain size, improving the density and shifting the Curie temperature. Under optimum sintering conditions and doping levels, the dielectric properties of pure $\mathrm{BaTiO}_{3}$ can be significantly enhanced with $\mathrm{Ta}_{2} \mathrm{O}_{5}$, $\mathrm{Nb}_{2} \mathrm{O}_{5}$ (Ahn et al., 2009). Judging the potential of pentavalent oxides used as dopants, the current study extensively examined the effects of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doping on the microstructure and dielectric properties of $\mathrm{BaTiO}_{3}$ ceramics. The ultimate aim of the study was to develop an environmentally-friendly ceramic capacitor for electronic industries.

## EXPERIMENTAL

Reagent-grade nano-sized oxide powder of $\mathrm{BaTiO}_{3}$ with a purity better than $99 \%$ was used as the starting raw materials. The powder was doped with $\mathrm{Ta}_{2} \mathrm{O}_{5}$ at concentrations ranging from $0.5-1.5$ mole $\%$. Both powders were mixed and milled for $18-20 \mathrm{hrs}$. Following that, the powders were dried and a binder PVA was added. Then the powders were pressed into pellets, approximately 5 mm thick and 12 mm in diameter, at a pressure of around 150 MPa using a hydraulic press (Figure 1(a)). Subsequently, the pressed samples were sintered in a high temperature furnace (Figure 1(b)). For a particular sintering cycle, the samples were at first heated to $550^{\circ} \mathrm{C}$ for one hour to remove the binder and were then heated to the desired sintering temperature. Single stage sintering was carried out in the range of $1250-1300^{\circ} \mathrm{C}$ for two hours. Whereas, two-stage sintering was carried out by keeping the 1st stage sintering condition fixed at $1320^{\circ} \mathrm{C}$ for zero hours and only varying the holding time during 2nd stage sintering from 4-6 hrs at a constant temperature of $1280^{\circ} \mathrm{C}$. After sintering, X-ray diffraction (XRD) was performed in order to determine the percent theoretical density (\%TD), phase and structural analysis. Moreover, micro-structural analysis was performed using a field emission scanning electron microscope (FESEM) (Figure 2(a)), and the temperature dependence of the dielectric constant $(k)$ was measured at various frequencies ( 10 to 500 kHz ) at a heating rate of $4^{\circ} \mathrm{C} / \mathrm{min}$ using an Impedance analyser (Figure 2(b)).


Figure 1. Experimental setup for sample preparation: (a) Hydraulic press for pelletizing;
(b) High temperature furnace for sintering.


Figure 2. Experimental setup for characterisation and dielectric property measurement: (a) FE-SEM for microstructural analysis; (b) Impedance analyser for dielectric property measurement.

## RESULTS AND DISCUSSION

Numerical data on the effect of single stage sintering on the percent theoretical density (\%TD), grain size and dielectric constant $(k)$ of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples are tabulated in Table 1. Table 1 shows that for a particular mole $\%$ of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doping, the percent theoretical density of the $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples increased with increasing temperature. However, at a certain temperature, an increase in the $\mathrm{Ta}_{2} \mathrm{O}_{5}$ mole $\%$ resulted in a lowering of percent theoretical density of the samples.

Table 1. Percent theoretical density (\%TD), grain size and dielectric constant ( $k$ ) of single stage sintered $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples

| $\begin{aligned} & \text { SI } \\ & \text { No } \end{aligned}$ | $\begin{gathered} \text { Sinterin } \\ \text { g rate } \\ \left({ }^{\circ} \mathrm{C} / \mathrm{min}\right) \end{gathered}$ | Sintering temperatur e $\left({ }^{0} \mathrm{C}\right)$ | Holdin g time <br> (Hours) | Cooling rate $\left({ }^{0} \mathrm{C} / \mathrm{min}\right.$ ) | $\begin{gathered} \text { Dopin } \\ \mathrm{g} \\ \text { mole } \\ \% \end{gathered}$ | $\begin{gathered} \% \mathrm{~T} \\ \mathrm{D} \end{gathered}$ | $\begin{gathered} \text { Grai } \\ \mathrm{n} \\ \text { size } \\ (\mu \mathrm{m}) \end{gathered}$ | Dielectric constant (k) at room temperatur e |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 5 | 1250 | 2hrs | 3 | 0.5 | 91.7 | 0.30 | 1700-2000 |
| 2 | 5 | 1250 | 2 hrs | 3 | 1.0 | 90.5 | 0.28 | 1700-2000 |
| 3 | 5 | 1250 | 2 hrs | 3 | 1.5 | 88.0 | 0.25 | 1700-2000 |
| 4 | 5 | 1275 | 2 hrs | 3 | 0.5 | 92.0 | 0.35 | 1700-2000 |
| 5 | 5 | 1275 | 2 hrs | 3 | 1.0 | 91.6 | 0.30 | 1700-2000 |
| 6 | 5 | 1275 | 2 hrs | 3 | 1.5 | 90.0 | 0.26 | 1700-2000 |
| 7 | 5 | 1300 | 2 hrs | 3 | 0.5 | 93.0 | 1.10 | 13000 |
| 8 | 5 | 1300 | 2 hrs | 3 | 1.0 | 92.0 | 0.40 | 1700-2000 |
| 9 | 5 | 1300 | 2 hrs | 3 | 1.5 | 91.5 | 0.38 | 1700-2000 |

It is also observed from Table 1 that for all $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples sintered within the temperature range of $1250-1275^{\circ} \mathrm{C}, 2 \mathrm{hrs}$ of holding time proved to be insufficient, resulting in samples with a fine grain size. Due to their fine grain size, all these samples exhibited a poor dielectric constant in spite of having a high percent theoretical density. However, at $1300^{\circ} \mathrm{C}$ the 2 hrs of the holding time proved to be sufficient for the 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample. As a result, the sample exhibited a high percent theoretical density, satisfactory grain size, and moderate dielectric constant. For $1.0-1.5$ mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples, the holding time again proved to be inadequate and resulted in samples of fine grain size. As a consequence, these samples again presented a poor dielectric constant. All these results are consistent with the findings of Manalert and Rahaman (1998), who concluded that the amount of pentavalent oxide dopants controls the microstructure of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples. For the penta-valance donor cations, the grain boundary mobility initially increases with cation concentration but then decreases significantly above a doping threshold of $0 \cdot 3-0.5$ mole $\%$. Generally dopants need very low to almost no energy to concentrate at the grain boundaries. However, energy is required to incorporate a dopant ion into an individual lattice site in complex oxides. The amount of energy required is related to the distortions, i.e. difference in ionic radii, and the formation of compensating defects during the incorporation of aliovalent ions that have different valence states (Ahn et al., 2009). Thus, more energy was required in our research for the diffusion of the dopants to move inwards from the grain boundary into the lattice and to reduce the pinning effect. The required energy could have been supplied by increasing the holding time of the single stage sintering. But this route was associated with a high cost and would make the project less viable for industrial application. As a result, further research was focused on two-stage sintering. Numerical data of the effect of two-stage sintering on percent theoretical density (\%TD), grain size and dielectric constant $(k)$ of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples are tabulated in Table 2.

Table 2. Percent theoretical density (\%TD), grain size and dielectric constant ( $k$ ) of twostage sintered $\mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples.

| $\begin{gathered} \text { SI } \\ \text { No: } \end{gathered}$ | Sintering rate ( ${ }^{0} \mathrm{C} / \mathrm{min}$ ) | Sintering temperature and holding time ( $1^{\text {st }}$ stage) | Sintering temperature and holding time (2 ${ }^{\text {nd }}$ stage) | Cooling rate ( ${ }^{0} \mathrm{C} / \mathrm{min}$ ) | Doping mole \% | \%TD | $\begin{gathered} \text { Grain } \\ \text { size } \\ (\mu \mathrm{m}) \end{gathered}$ | Dielectric constant (k) at room temperature |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 4 \mathrm{hrs} \end{gathered}$ | 3 | 0.5 | 95.0 | 1.80 | 9500 |
| 2 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 4 \mathrm{hrs} \end{gathered}$ | 3 | 1.0 | 93.0 | 1.30 | 16500 |
| 3 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 4 \mathrm{hrs} \end{gathered}$ | 3 | 1.5 | 91.0 | 1.10 | 17900 |
| 4 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 6 \mathrm{hrs} \end{gathered}$ | 3 | 0.5 | 93.5 | 1.90 | 8900 |
| 5 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 6 \mathrm{hrs} \end{gathered}$ | 3 | 1.0 | 92.1 | 1.40 | 13700 |
| 6 | 5 | $\begin{gathered} 1320^{\circ} \mathrm{C} \text { for } \\ 0 \mathrm{hrs} \end{gathered}$ | $\begin{gathered} 1280^{\circ} \mathrm{C} \text { for } \\ 6 \mathrm{hrs} \end{gathered}$ | 3 | 1.5 | 92.9 | 1.20 | 19000 |

Table 2 shows that for a 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample, 4 hrs of holding time during $2^{\text {nd }}$ stage sintering decreased the dielectric constant value due to an excessive increase in grain size. The high grain growth of the sample may have resulted from the complete diffusion of $\mathrm{Ta}_{2} \mathrm{O}_{5}$ into the bulk material. This phenomenon is clearly evident from the SEM micrograph of Figure 3(a), which shows no evidence of bimodal grain size distribution or the pinning effect. However, for a 1.0 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample, 4hrs of holding time produced controlled grain growth and increased the dielectric constant value to a moderate level, which is shown in the SEM micrograph of Figure 3(b). For the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample, 4 hrs of holding time might not have been sufficient for complete diffusion of the dopants into the lattice, and as a result there was evidence of the pinning effect. The SEM micrograph of Figure 3(c) clearly shows the presence of bimodal grain size distribution. However, the combination of a moderate percent theoretical density and suitable grain size of 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample resulted in the best dielectric properties for 4 hrs of holding time.


Figure 3. SEM micrographs (X 50,000) of: (a) 0.5 mole \%; (b) 1.0 mole \%; and (c) 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples sintered at $1320^{\circ} \mathrm{C}(0 \mathrm{hrs})$ and $1280^{\circ} \mathrm{C}(4 \mathrm{hrs})$.

Table 2 also shows that for 6 hrs of holding time, the 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample showed even more exaggerated grain growth in comparison to the sample sintered at 4 hrs of holding time (Figure 4(a)). This excessive grain growth resulted in a decrease in percent theoretical density of the sample and also lowered its dielectric constant. However, under the same sintering conditions, the 1.0 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample showed controlled grain growth but larger than the sample sintered under previous sintering conditions (Figure $4(\mathrm{~b})$ ). For the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample, even 6 hrs of holding time showed signs of the pinning effect. But the grain size of the sample was slightly higher than the sample sintered at 4 hrs of holding time (Figure 4(c)). Due to the combination of high percent theoretical density and optimum grain size, the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample exhibited the best dielectric properties for 6hrs of holding time.


Figure 4. SEM micrographs (X 50,000) of: (a) 0.5 mole \%; (b) 1.0 mole \%; and (c) 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples sintered at $1320^{\circ} \mathrm{C}(0 \mathrm{hrs})$ and $1280^{\circ} \mathrm{C}(6 \mathrm{hrs})$.

The XRD patterns of Figure 5 indicate the formation of both tetragonal and cubic phases of $\mathrm{BaTiO}_{3}$. The XRD pattern for the 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 4 hrs shows the presence of twin peaks
similar to pure $\mathrm{BaTiO}_{3}$ powder (Figures $5(\mathrm{a})$ and $5(\mathrm{~b})$ ). So it can be concluded that for 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doping, the $\mathrm{BaTiO}_{3}$ retained the tetragonal perovskite structure. However, the XRD pattern for the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 4 hrs shows no evidence of twin peaks, which indicates that 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ reduced the tetragonality and stabilised the cubic phase of $\mathrm{BaTiO}_{3}$ (Figure 5 (c)). This phenomenon is also consistent in the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ samples sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 6 hrs (Figure $5(\mathrm{~d})$ ). So it can be stated that 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ stabilised the cubic phase of $\mathrm{BaTiO}_{3}$.


Figure 5. XRD plots for: (a) Pure Barium Titanate powder; (b) 0.5 mole \%; and (c) 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}(0 \mathrm{hrs})$ and $1280^{\circ} \mathrm{C}(4 \mathrm{hrs})$; and (d) 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}$ ( 0 hrs ) and $1280^{\circ} \mathrm{C}$ ( 6 hrs ).

The findings from the XRD patterns are also consistent with the results of the temperature dependence of the dielectric constant $(k)$. Although the Curie temperature $\left(T_{c}\right)$ for pure $\mathrm{BaTiO}_{3}$ is $120^{\circ} \mathrm{C}$, the Curie point of the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 4 hrs shifted to $84^{\circ} \mathrm{C}$ (Figure 6), and the Curie point of the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 6 hrs shifted to $80^{\circ} \mathrm{C}$ (Figure 7). According to theory, at a fixed temperature an increase in frequency results in a decrease in dielectric constant (Bowen et al., 1976; Carter and Norton, 2007; Richerson, 1992). This effect of tfrequency on the dielectric constant is clearly evident in Figure 8. The best stable value of dielectric constant as a function of temperature was obtained around 18000 for the 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sample sintered at $1320^{\circ} \mathrm{C}$ for 0 hrs and $1280^{\circ} \mathrm{C}$ for 6hrs (Figure 7).


Figure 6. Variation in dielectric constant with temperature 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}(0 \mathrm{hrs})$ and $1280^{\circ} \mathrm{C}(4 \mathrm{hrs})$.


Figure 7. Variation in dielectric constant with temperature of 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doped $\mathrm{BaTiO}_{3}$ sintered at $1320^{\circ} \mathrm{C}(0 \mathrm{hrs})$ and $1280^{\circ} \mathrm{C}(6 \mathrm{hrs})$.


Figure 8. Room temperature dielectric constant of samples sintered at optimum sintering cycles.

## CONCLUSION

An average grain size of around $1 \mu \mathrm{~m}$ normally provides a good condition for the dielectric properties of doped $\mathrm{BaTiO}_{3}$. In this research, grain sizes in the range of $0.8-$ $1.2 \mu \mathrm{~m}$ showed high values of dielectric constant, while a further increase in grain size deteriorated this property. It can also be concluded that up to 0.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ doping, $\mathrm{BaTiO}_{3}$ retained its tetragonal perovskite structure. However, 1.5 mole $\% \mathrm{Ta}_{2} \mathrm{O}_{5}$ stabilised the cubic phase of $\mathrm{BaTiO}_{3}$ and shifted the Curie temperature towards room temperature.

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